être l'alcaloïde le plus typique des feuilles des représentants du genre Hunteria. Les diverses espèces se distinguent assez nettement par la nature des autres alcaloïdes présents dans les feuilles H. corymbosa [9], H. umbellata [10, 11] et H. eburnea [3].

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## THE CO-OCCURRENCE OF RUTACRIDONE AND NORACRONYCINE IN THE ROOTS OF BOENNINGHAUSENIA ALBIFLORA

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**Key Word Index**—Boenninghausenia albiflora; Rutaceae; acridone alkaloids; rutacridone; 1-hydroxyacridone; 1-hydroxy-N-methylacridone; 1,7-dihydroxy-N-methylacridone; acridone glycoside.

TLC analysis of the methanolic extracts of Boenninghausenia albiflora Reichb. revealed the presence of several acridone alkaloids in the roots [1, 2]. Since no acridone alkaloids have been reported from this plant before, solvent-solvent fractionation was undertaken for the separation of these compounds [3]. In a preliminary communication [4] the isolation of three of these yellow alkaloids was described. One of them  $(Br_1)$  was identified as 1-hydroxy-N-methylacridone (5), the two, remaining one were tentatively designated  $Br_2$  and  $Br_4$ . In the present paper we report on the characterisation of these two compounds and on three further compounds isolated from the same extract. Two of them  $(Br_5)$  and  $Br_6$ ) were isolated from the benzene soluble fraction of the extract (see Experimental), the third one  $(Br_7)$  was separated from the ethyl acetate soluble fraction.

Two  $(Br_5$  and  $Br_6)$  were identified by direct comparison with authentic samples as rutacridone (1) and noracronycine (2) respectively. The alkaloid  $Br_2$ ,  $C_{13}H_9O_2N$  (M<sup>+</sup> at m/e 211), in its NMR spectrum indicates the presence of a chelated C-1-OH group and of seven aromatic protons. This region shows a striking similarity to that of N-methylacridone. Thus structure 4 (1-hydroxyacridone) can be formulated and this is confirmed by the conversion of 4 to 1-hydroxy-N-methylacridone (5) via methylation.

The molecular ion of  $Br_4$  (m/e 241) indicates a molecular formula  $C_{14}H_{11}O_3N$ . The NMR spectrum exhibits two singlets at 3.93 and 14 (C-1-OH) ppm. The assignment of the first mentioned signal to a methylimino-group is deducible from its downfield shift in

trifluoroacetic acid. The brown colour reaction of  $Br_4$  with alcoholic ferric chloride [4] is characteristic for a free phenolic OH-group. An ABX-system analogous to those of  $Br_2$  and  $Br_3$  shows three adjacent protons in position 2, 3 and 4. The remaining signals of the other three protons must be assigned to the second benzene ring. On this basis, structure 6 is favoured. This structural proposal was confirmed by synthesis [5].

(6): Me

(4): R = H; (5): R = Me

Compound  $Br_7$  is slightly water-soluble. Its IR spectrum exhibits very intense OH bands at 3400 and  $1060 \text{ cm}^{-1}$  indicating its polyhydroxylated nature. Acidic

hydrolysis of  $Br_7$  yielded glucose and a yet unknown acridone aglycone. Thus compound  $Br_7$  must be an acridone glucoside.

To our knowledge, this is the first report on the isolation of 1-hydroxyacridone and of 1.7-dihydroxy-N-methylacridone (6) from a plant species; compound  $Br_7$  is probably a new natural acridone alkaloid, too.

The most interesting feature of our results is the joint occurrences of rutacridone (1) and noracronycine (2) in the same plant. To our knowledge Boenninghausenia albiflora is the second rutaceous species known to contain both furanoid and pyranoid derivatives in the acridone series. The only other species containing both types is Atlantia ceylanica; either furanoid or pyranoid acridones occur in other species, e.g. Acronychia baueri, Glycosmis pentaphylla, Atalantia monophylla and Ruta graveolens [6-8].

## **EXPERIMENTAL**

Plant material was collected in May 1974 in the vicinity of Pankhasari, India. The mps of the compounds are uncorr. For the chromatographic separation of the components 70–230 mesh Si gel, Merck TLC Si gel and polyamide were used.

Isolation. 1100 g dried root material was extracted at room temp. with MeOH. The MeOH extract was concd in vacuo to 500 ml, diluted with equal vol. of  $H_2O$  and extracted with  $C_6H_6$  and EtOAc successively. After evapn of the solvents both fractions were chromatographed on polyamide. The  $C_6H_6$  fraction, when eluted with MeOH- $H_2O$  (3:2) afforded an acridone mixture consisting of three major components  $(Br_2, Br_3, Br_4)$  They were further separated on preparative Si gel TLC ( $C_6H_6$ ,  $C_6H_6$ -EtOAc.19:1 and 4:1 systems) then on Sephadex LH 20 columns

Br<sub>2</sub> = l-hydroxyacridone (4). Mp 280° (subl. decom.), yellow needles from acetone. UV<sub>max</sub> (MeOH): 241 (infl.) 255 (infl.), 260, 303, 314, 400 nm IRv<sub>max</sub> (KBr): 3100, 2950, 1640, 1600, 1530, 1470, 1280, 1230, 930, 770 cm<sup>-1</sup>, see also [4]. NMR ((CD<sub>3</sub>)<sub>2</sub>CO): 14.0 (1Hs; C-1-OH), 8,3 (1H q, H-8), 7.8-7.1 (4H complex m, H-3, H-5, H-6 a. H-7), 6,9 and 6.5 ppm (each 1H q (f<sub>A,B</sub> = 8 c/s) H-4, H-2). MS: m/e 211 (100) (M<sup>+</sup>), 183 (29). 154 (31), 127 (12.5). 105 5 (4), 91.5 (14), 77 (31). 1 mg 3 was dissolved in dry Me<sub>2</sub>CO and added to a mixture of dry K<sub>2</sub>CO<sub>3</sub> in MeI. The mixture was kept at room temp for one day, then filtered and evapd. The product was identified as 1-hydroxy-f<sub>methylacridone</sub> (5) on Si gel and polyamide TLC Systems: C<sub>6</sub>H<sub>6</sub>-EtOAc (4:1): toluene-EtOAc-HCO<sub>2</sub>H (5:4.1): MeOH-H<sub>2</sub>O (3:1).

Br<sub>3</sub> = 1-hydroxy-N-methylacridone 5. mp, mmp, UV and IR data see [4] MS: m/e 225 (100, M<sup>+</sup>), 210 (3), 197 (7.5), 196 (7), 182 (16), 168 (3.5), 167 (3), 154 (5), 127 (4), 112.5 (2), 98 5, 77 (6).

 $Br_4=\it{1.7-dihydroxy-N-methylacridone}$  (6) mp 308° (subl., decom.) yellow needles from acetone. UV  $_{max}$  (MeOH): 244 (infl.), 262, 272, 314 (infl.), 328 (infl.), 410 nm  $IRv_{max}$  (KBr):

3250, 2900, 1635, 1590, 1520, 1480, 1280, 1230, 810, 765 cm<sup>-1</sup>, see also [4]. NMR ((CD<sub>3</sub>)<sub>2</sub>CO)· 14.0 (1H s; C-1-OH), 7.75 (1H d (J = 2 c/s; H-8), 7-7.5 (3H, complex m, H-3, H-5, H-6), 6.9 (1H q,( $J_{A,B}$  = 8 c/s) H-4), 6.5 (1H q, ( $J_{A,B}$  = 8 c/s), H-2) and 3.93 ppm (3H s, Me—N $\searrow$ , in trifluoroacetic acid 4.10 ppm) MS: m/e 241 (100, M $^+$ ), 226 (67), 212 (8.5), 193 (15), 183 (2), 170 (18), 154 (3.5), 142 (9), 141 (8.5), 140 (8), 120.5 (19), 115 (11.5), 85 (7.5), 77 (9) The fractions eluted with MeOH–H<sub>2</sub>O (4:1) contained two yellow compounds ( $Br_5$  and  $Br_6$ ). These substances were separated on PLC ( $C_6H_6$ –EtOAc (19:1)) and were further purified on Sephadex LH 20 column [4].

 $B_{15} = rutacridone$  (1). mp 162–164 (lit 161 162) mmp 162–164° yellow needles from Me<sub>2</sub>CO-petrol Co-chromatography on Si gel and polyamide layers gave identical  $R_f$ -values with those of authentic rutacridone. MS:  $m_e$  307 (100, M<sup>+</sup>), 292 (30), 278 (29), 264 (27), 250 (16), 239 (16), 236 (15), 222 (6), 208 (12), 183 (4), 180 (6.5), 167 (5), 154 (5), 146 (2), 140 (5), 127 (1), 115 (1,2), 107 (9), 89 (3, 5), 77 (13).

Br<sub>6</sub> = noracronycine **2**. mp 200–204°. (ltt. 198–200°), mmp 203–205° orange crystals from  $C_6H_6$ . UV  $\lambda_{max}$  250, 260 (sh), 275, 301, 335 (sh), 405 nm. IR  $v_{max}$  3400, 2930, 1620, 1590, 1500, 1410, 1060 cm<sup>-1</sup>. Co-chromatography on Si gel and polyamide TLC gave identical  $R_f$ -values with those of the authentic noracronycine. 0.5 mg **2** was dissolved in dry Me<sub>2</sub>CO and added to a soln of dry K<sub>2</sub>CO<sub>3</sub> in Me<sub>2</sub>SO<sub>4</sub>. The mixture was kept at room temp. for one week. The product was identified as acronycine **3** on Si gel and polyamid TLC.

 ${\rm Br}_7=acridon\text{-}glucoside}$  Mp  $167\text{-}177^\circ$  amorphous powder from EtOAc-MeOH 17 mg of  $Br_7$  was refluxed on a stream bath in NH<sub>2</sub>SO<sub>4</sub> for 1 hr. The mixture was then neutralized with BaCO<sub>3</sub>, filtered, washed and extracted with EtOAc. In the neutralized aq. layer glucose was identified by PC. 13 mg  $Br_7$  were acetylated at room temp. with  ${\rm Ac}_2{\rm O-Py}$ . The acetate was obtained as yellow crystals, mp 160-165.

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